



National  
Qualifications  
2016

**X757/77/11**

**Physics  
Relationships Sheet**

TUESDAY, 24 MAY  
9:00 AM – 11:30 AM



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## Relationships required for Physics Advanced Higher

$$v = \frac{ds}{dt}$$

$$a = \frac{dv}{dt} = \frac{d^2s}{dt^2}$$

$$v = u + at$$

$$s = ut + \frac{1}{2}at^2$$

$$v^2 = u^2 + 2as$$

$$\omega = \frac{d\theta}{dt}$$

$$\alpha = \frac{d\omega}{dt} = \frac{d^2\theta}{dt^2}$$

$$\omega = \omega_o + \alpha t$$

$$\theta = \omega_o t + \frac{1}{2}\alpha t^2$$

$$\omega^2 = \omega_o^2 + 2\alpha\theta$$

$$s = r\theta$$

$$v = r\omega$$

$$a_t = r\alpha$$

$$a_r = \frac{v^2}{r} = r\omega^2$$

$$F = \frac{mv^2}{r} = mr\omega^2$$

$$T = Fr$$

$$T = I\alpha$$

$$L = mvr = mr^2\omega$$

$$L = I\omega$$

$$E_K = \frac{1}{2}I\omega^2$$

$$F = G \frac{Mm}{r^2}$$

$$V = -\frac{GM}{r}$$

$$v = \sqrt{\frac{2GM}{r}}$$

$$\text{apparent brightness, } b = \frac{L}{4\pi r^2}$$

$$\text{Power per unit area} = \sigma T^4$$

$$L = 4\pi r^2 \sigma T^4$$

$$r_{\text{Schwarzschild}} = \frac{2GM}{c^2}$$

$$E = hf$$

$$\lambda = \frac{h}{p}$$

$$mvr = \frac{nh}{2\pi}$$

$$\Delta x \Delta p_x \geq \frac{h}{4\pi}$$

$$\Delta E \Delta t \geq \frac{h}{4\pi}$$

$$F = qvB$$

$$\omega = 2\pi f$$

$$a = \frac{d^2y}{dt^2} = -\omega^2 y$$

$$y = A \cos \omega t \quad \text{or} \quad y = A \sin \omega t$$

$$v = \pm \omega \sqrt{(A^2 - y^2)}$$

$$E_K = \frac{1}{2} m \omega^2 (A^2 - y^2)$$

$$E_P = \frac{1}{2} m \omega^2 y^2$$

$$y = A \sin 2\pi \left( ft - \frac{x}{\lambda} \right)$$

$$E = kA^2$$

$$\phi = \frac{2\pi x}{\lambda}$$

$$\text{optical path difference} = m\lambda \quad \text{or} \quad \left( m + \frac{1}{2} \right) \lambda$$

where  $m = 0, 1, 2, \dots$

$$\Delta x = \frac{\lambda}{2d}$$

$$d = \frac{\lambda}{4n}$$

$$\Delta x = \frac{\lambda D}{d}$$

$$n = \tan i_p$$

$$F = \frac{Q_1 Q_2}{4\pi \epsilon_0 r^2}$$

$$E = \frac{Q}{4\pi \epsilon_0 r^2}$$

$$V = \frac{Q}{4\pi \epsilon_0 r}$$

$$F = QE$$

$$V = Ed$$

$$F = IlB \sin \theta$$

$$B = \frac{\mu_0 I}{2\pi r}$$

$$c = \frac{1}{\sqrt{\epsilon_0 \mu_0}}$$

$$t = RC$$

$$X_C = \frac{V}{I}$$

$$X_C = \frac{1}{2\pi fC}$$

$$\mathcal{E} = -L \frac{dI}{dt}$$

$$E = \frac{1}{2} LI^2$$

$$X_L = \frac{V}{I}$$

$$X_L = 2\pi fL$$

$$\frac{\Delta W}{W} = \sqrt{\left( \frac{\Delta X}{X} \right)^2 + \left( \frac{\Delta Y}{Y} \right)^2 + \left( \frac{\Delta Z}{Z} \right)^2}$$

$$\Delta W = \sqrt{\Delta X^2 + \Delta Y^2 + \Delta Z^2}$$

$$d = \bar{v}t$$

$$s = \bar{v}t$$

$$v = u + at$$

$$s = ut + \frac{1}{2}at^2$$

$$v^2 = u^2 + 2as$$

$$s = \frac{1}{2}(u + v)t$$

$$W = mg$$

$$F = ma$$

$$E_w = Fd$$

$$E_p = mgh$$

$$E_k = \frac{1}{2}mv^2$$

$$P = \frac{E}{t}$$

$$p = mv$$

$$Ft = mv - mu$$

$$F = G \frac{Mm}{r^2}$$

$$t' = \frac{t}{\sqrt{1 - \left(\frac{v}{c}\right)^2}}$$

$$l' = l\sqrt{1 - \left(\frac{v}{c}\right)^2}$$

$$f_o = f_s \left( \frac{v}{v \pm v_s} \right)$$

$$z = \frac{\lambda_{observed} - \lambda_{rest}}{\lambda_{rest}}$$

$$z = \frac{v}{c}$$

$$v = H_0d$$

$$E_w = QV$$

$$E = mc^2$$

$$E = hf$$

$$E_k = hf - hf_0$$

$$E_2 - E_1 = hf$$

$$T = \frac{1}{f}$$

$$v = f\lambda$$

$$d\sin\theta = m\lambda$$

$$n = \frac{\sin\theta_1}{\sin\theta_2}$$

$$\frac{\sin\theta_1}{\sin\theta_2} = \frac{\lambda_1}{\lambda_2} = \frac{v_1}{v_2}$$

$$\sin\theta_c = \frac{1}{n}$$

$$I = \frac{k}{d^2}$$

$$I = \frac{P}{A}$$

$$\text{path difference} = m\lambda \quad \text{or} \quad \left(m + \frac{1}{2}\right)\lambda \quad \text{where } m = 0, 1, 2, \dots$$

$$\text{random uncertainty} = \frac{\text{max. value} - \text{min. value}}{\text{number of values}}$$

$$V_{peak} = \sqrt{2}V_{rms}$$

$$I_{peak} = \sqrt{2}I_{rms}$$

$$Q = It$$

$$V = IR$$

$$P = IV = I^2R = \frac{V^2}{R}$$

$$R_T = R_1 + R_2 + \dots$$

$$\frac{1}{R_T} = \frac{1}{R_1} + \frac{1}{R_2} + \dots$$

$$E = V + Ir$$

$$V_1 = \left( \frac{R_1}{R_1 + R_2} \right) V_s$$

$$\frac{V_1}{V_2} = \frac{R_1}{R_2}$$

$$C = \frac{Q}{V}$$

$$E = \frac{1}{2}QV = \frac{1}{2}CV^2 = \frac{1}{2} \frac{Q^2}{C}$$

## Additional Relationships

### Circle

$$\text{circumference} = 2\pi r$$

$$\text{area} = \pi r^2$$

### Sphere

$$\text{area} = 4\pi r^2$$

$$\text{volume} = \frac{4}{3}\pi r^3$$

### Trigonometry

$$\sin \theta = \frac{\textit{opposite}}{\textit{hypotenuse}}$$

$$\cos \theta = \frac{\textit{adjacent}}{\textit{hypotenuse}}$$

$$\tan \theta = \frac{\textit{opposite}}{\textit{adjacent}}$$

$$\sin^2 \theta + \cos^2 \theta = 1$$

### Moment of inertia

point mass

$$I = mr^2$$

rod about centre

$$I = \frac{1}{12}ml^2$$

rod about end

$$I = \frac{1}{3}ml^2$$

disc about centre

$$I = \frac{1}{2}mr^2$$

sphere about centre

$$I = \frac{2}{5}mr^2$$

### Table of standard derivatives

$f(x)$	$f'(x)$
$\sin ax$	$a \cos ax$
$\cos ax$	$-a \sin ax$

### Table of standard integrals

$f(x)$	$\int f(x)dx$
$\sin ax$	$-\frac{1}{a} \cos ax + C$
$\cos ax$	$\frac{1}{a} \sin ax + C$

# Electron Arrangements of Elements

Group 1    Group 2    Group 3    Group 4    Group 5    Group 6    Group 7    Group 0

(18)

(13)	5	<b>B</b>	2, 3	<b>C</b>	2, 4	<b>N</b>	2, 5	<b>O</b>	2, 6	<b>F</b>	2, 7	<b>Ne</b>	2, 8	<b>He</b>	2	<b>Helium</b>
		Boron		Carbon		Nitrogen		Oxygen		Fluorine		Neon				
	13	<b>Al</b>	2, 8, 3	<b>Si</b>	2, 8, 4	<b>P</b>	2, 8, 5	<b>S</b>	2, 8, 6	<b>Cl</b>	2, 8, 7	<b>Ar</b>	2, 8, 8			
		Aluminium		Silicon		Phosphorus		Sulphur		Chlorine		Argon				
	31	<b>Ga</b>	2, 8, 18, 3	<b>Ge</b>	2, 8, 18, 4	<b>As</b>	2, 8, 18, 5	<b>Se</b>	2, 8, 18, 6	<b>Br</b>	2, 8, 18, 7	<b>Kr</b>	2, 8, 18, 8			
		Gallium		Germanium		Arsenic		Selenium		Bromine		Krypton				
	49	<b>In</b>	2, 8, 18, 18, 3	<b>Sn</b>	2, 8, 18, 18, 4	<b>Sb</b>	2, 8, 18, 18, 5	<b>Te</b>	2, 8, 18, 18, 6	<b>I</b>	2, 8, 18, 18, 7	<b>Xe</b>	2, 8, 18, 18, 8			
		Indium		Tin		Antimony		Tellurium		Iodine		Xenon				
	81	<b>Tl</b>	2, 8, 18, 32, 18, 3	<b>Pb</b>	2, 8, 18, 32, 18, 4	<b>Bi</b>	2, 8, 18, 32, 18, 5	<b>Po</b>	2, 8, 18, 32, 18, 6	<b>At</b>	2, 8, 18, 32, 18, 7	<b>Rn</b>	2, 8, 18, 32, 18, 8			
		Thallium		Lead		Bismuth		Polonium		Astatine		Radon				

(12)	30	<b>Zn</b>	2, 8, 18, 2	<b>Cu</b>	2, 8, 18, 1	<b>Ni</b>	2, 8, 16, 2	<b>Co</b>	2, 8, 15, 2	<b>Rh</b>	2, 8, 18, 16, 1	<b>Pd</b>	2, 8, 18, 18, 0	<b>Ag</b>	2, 8, 18, 18, 1	<b>Cd</b>	2, 8, 18, 18, 2	<b>Hg</b>	2, 8, 18, 32, 18, 2	<b>Mt</b>	2, 8, 18, 32, 32, 15, 2	<b>Mt</b>	109		
		Zinc		Copper		Nickel		Cobalt		Rhodium		Palladium		Silver		Cadmium		Mercury		Mt		Mt		Mt	
	48	<b>Cd</b>	2, 8, 18, 18, 2	<b>Ag</b>	2, 8, 18, 18, 1	<b>Pt</b>	2, 8, 18, 32, 17, 1	<b>Au</b>	2, 8, 18, 32, 18, 1	<b>Hg</b>	2, 8, 18, 32, 18, 2	<b>Ir</b>	2, 8, 18, 32, 15, 2	<b>Rh</b>	2, 8, 18, 16, 1	<b>Pd</b>	2, 8, 18, 18, 0	<b>Ag</b>	2, 8, 18, 18, 1	<b>Cd</b>	2, 8, 18, 18, 2	<b>Hs</b>	2, 8, 18, 32, 32, 14, 2	<b>Hs</b>	108

(1)	1	<b>H</b>	1	<b>He</b>	2
		Hydrogen		Helium	
	3	<b>Li</b>	2, 1	<b>Be</b>	2, 2
		Lithium		Beryllium	
	11	<b>Na</b>	2, 8, 1	<b>Mg</b>	2, 8, 2
		Sodium		Magnesium	
	19	<b>K</b>	2, 8, 8, 1	<b>Ca</b>	2, 8, 8, 2
		Potassium		Calcium	
	37	<b>Rb</b>	2, 8, 18, 8, 1	<b>Sr</b>	2, 8, 18, 8, 2
		Rubidium		Strontium	
	55	<b>Cs</b>	2, 8, 18, 18, 8, 1	<b>Ba</b>	2, 8, 18, 18, 8, 2
		Caesium		Barium	
	87	<b>Fr</b>	2, 8, 18, 32, 18, 8, 1	<b>Ra</b>	2, 8, 18, 32, 18, 8, 2
		Francium		Radium	

Atomic number

Symbol

Electron arrangement

Name

Key

## Transition Elements

(16)	69	<b>Tm</b>	2, 8, 18, 31, 8, 2	<b>Lu</b>	2, 8, 18, 32, 9, 2	<b>Lu</b>	2, 8, 18, 32, 9, 2
		Thulium		Lutetium			
	70	<b>Yb</b>	2, 8, 18, 32, 8, 2	<b>No</b>	2, 8, 18, 32, 8, 2	<b>No</b>	2, 8, 18, 32, 8, 2
		Ytterbium		Nobelium			
	71	<b>Lu</b>	2, 8, 18, 31, 8, 2	<b>Lr</b>	2, 8, 18, 32, 8, 2	<b>Lr</b>	2, 8, 18, 32, 8, 2
		Lutetium		Lawrencium			

## Lanthanides

## Actinides

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